



General information	
Academic subject	<b>Physical Methods in Organic Chemistry</b>
Degree course	CTF
Year of study	III
European Credit Transfer and Accumulation System (ECTS)	8 CFU
Language	Italian
Academic Year	2021/2022
Academic calendar (starting and ending date)	March 2022 – June 2022
Attendance	YES

Professor/ Lecturer	
Name and Surname	<b>Renzo LUISI</b>
E-mail	<b>renzo.luisi@uniba.it</b>
Telephone	<b>+39-0805442762</b>
Department and address	<b>Dept. of Pharmacy – Drug Sciences</b>
Virtual headquarters	<b>Teams platform</b>
Tutoring (time and day)	<b>Monday – Friday, 15:00 – 17:00</b>

Syllabus	
<b>Learning Objectives</b>	By a combination of lectures, exercises and individual semester assignment, the principles of UV/VIS-;IR-;1H- and 13C-NMR (1D and 2D spectra); and mass spectrometry are discussed. The course is especially focused on analysing and interpreting spectral data of organic compounds. Spectroscopic determination of the structure of organic molecules.
<b>Course prerequisites</b>	Basic knowledge of organic chemistry and physics
<b>Contents</b>	Course layout Overview of electromagnetic spectrum and its application to the study of chemical molecules UV-Vis. <ul style="list-style-type: none"><li>• Principle of UV-Vis spectroscopy.</li><li>• Interpretation of UV-Vis spectroscopy.</li><li>• Electronic transitions in organic molecules, selection rules, application of Beer Lambert law, qualitative and quantitative analysis by UV-Vis spectroscopy.</li><li>• Instrumentation and Application of UV-Vis spectroscopy.</li></ul> IR. <ul style="list-style-type: none"><li>• Principle Infra-Red Spectroscopy.</li><li>• Basic concepts, experimental methods, functional group analysis and identification using IR spectroscopy, structural effects on vibrational frequency.</li><li>• Interpretation of Infra-Red Spectroscopy.</li><li>• Instrumentation and Application of Infra-Red Spectroscopy</li></ul> NMR <ul style="list-style-type: none"><li>• Principles of Nuclear Magnetic Resonance Spectroscopy.</li><li>• Spin ½ nuclei, 1H and 13C-NMR spectroscopy, FT-NMR method. Chemical shifts, spin-spin coupling, spin-spin splitting pattern recognition for structure elucidation, coupling constants.</li><li>• Interpretation of Nuclear Magnetic Resonance Spectroscopy.</li><li>• Second order effects in NMR spectrum, AB and AA'BB', ABC spin systems.</li><li>• Solving simple structure elucidation problems with 1H and 13C NMR spectroscopy.</li></ul>



	<ul style="list-style-type: none"><li>• Stereochemistry determination using NMR techniques. Study of dynamic processes by NMR spectroscopy.</li><li>• Various ionization methods – EI, CI, ESI and MALDI methods, fragmentation patterns of simple organic molecules, Use of HRMS.</li><li>• Fragmentation patterns of simple organic molecules (continued), solving structure elucidation problems using mass spectrometry.</li><li>• Mass spectrometry and the combined use of spectroscopic techniques for structure elucidation.</li><li>• Instrumentation and Application of Nuclear Magnetic Resonance Spectroscopy MS spectrometry</li><li>• Solving structure elucidation problems using multiple spectroscopic data (NMR, MS, IR and UV-Vis).</li></ul>
<b>Books and bibliography</b>	R. M. SILVERSTEIN, F. X. WEBSTER- Identificazione Spettroscopica di Composti Organici- Casa Editrice Ambrosiana. M. HESSE, H. MEIER, B. ZEEH - Metodi Spettroscopici nella Chimica Organica- EdiSES WHITTAKER D. - Interpretation of Organic Spectra - RSC. E. PRETSCH, P. BUHLMANN, C. AFFOLTER - Structure Determination of Organic Compounds Table of spectral data - Springer. D. L. PAVIA, G. M. LAMPMAN E G. S. KRIZ - Introduction to Spectroscopy: a Guide for Students of Organic Chemistry - Saunders Golden Sunburst Series - Saunders College Publishing - Philadelphia. London - Toronto.
<b>Additional materials</b>	Additional files will be provided to all students.

<b>Work schedule</b>			
Total	Lectures	Hands on (Laboratory, working groups, seminars, field trips)	Out-of-class study hours/ Self-study hours
<b>Hours</b>			
200	70	10	120
<b>ECTS</b>			
8	70	10	
<b>Teaching strategy</b>			
Classroom lectures, case studies, group problem solving. Lectures (4 hours per week), practical exercises (1 hours per week). E-learning not allowed.			
<b>Expected learning outcomes</b>			
<b>Knowledge and understanding on:</b>	After having completed the course, the student should be able to: <ul style="list-style-type: none"><li>○ Explain principles of UV, IR, MS and NMR spectroscopy.</li><li>○ Explain construction/design of spectrometers.</li><li>○ Operate UV and IR spectrometers, execute runs, acquire, and process spectra.</li><li>○ Analyse and interpret spectral data.</li><li>○ Use spectral data to elucidate an unknown structure or solve a structure related problem.</li><li>○ <i>Design the most suitable set of spectroscopic methods for a given structure-related problem.</i></li></ul>		
<b>Applying knowledge and understanding on:</b>	<ul style="list-style-type: none"><li>○ Spectroscopy and instrumentation</li><li>○ Method to approach structure elucidation</li><li>○ Capacity to analyse organic samples</li></ul>		



	<ul style="list-style-type: none"><li>○ Capacity to assign structure to organic molecules</li></ul>
<b>Soft skills</b>	<ul style="list-style-type: none"><li>● <i>Making informed judgments and choices</i><ul style="list-style-type: none"><li>○ Appropriate technical language.</li><li>○ Capacity to elaborate scientific data</li></ul></li><li>● <i>Communicating knowledge and understanding</i><ul style="list-style-type: none"><li>○ Capacity to transfer technical information</li></ul></li><li>● <i>Capacities to continue learning</i><ul style="list-style-type: none"><li>○ Capacity to expand knowledge in the field of chemistry and spectroscopy.</li></ul></li></ul>
<b>Assessment and feedback</b>	
Methods of assessment	<p><i>Compulsory exam assignment (written report): structure elucidation of an unknown compound by spectroscopic methods. An approved individual report counts for 50% of the exam. An oral assessment follows the written assignment.</i></p> <ul style="list-style-type: none"><li>- <i>Written exam duration (150 min)</i></li><li>- <i>Supplementary material allowed for the written exam.</i></li><li>- <i>Results published on the esse3 platform</i></li><li>- <i>Exercises available for all students</i></li></ul>
Evaluation criteria	<ul style="list-style-type: none"><li>● <i>Knowledge and understanding</i><ul style="list-style-type: none"><li>○ Appropriateness of the knowledge.</li><li>○ Capacity to use spectroscopic techniques.</li></ul></li><li>● <i>Applying knowledge and understanding</i><ul style="list-style-type: none"><li>○ Critical reasoning and correct technical terminology</li></ul></li><li>● <i>Autonomy of judgment</i><ul style="list-style-type: none"><li>○ Capacity to self-assess the correct sequence in a structural analysis.</li></ul></li><li>● <i>Communication skills</i><ul style="list-style-type: none"><li>○ Clarity of the exposition</li></ul></li><li>● <i>Capacities to continue learning</i><ul style="list-style-type: none"><li>○ Acquisition of advanced knowledge of spectroscopy, which are propaedeutic for the study of Medicinal Chemistry and for the synthesis of pharmaceutically relevant compounds</li></ul></li></ul>
Criteria for assessment and attribution of the final mark	<p><i>Mark is minimum 18 out of 30 and maximum 30 out of 30.</i></p> <p><i>50% of the maximum mark score is given to the written exam, and 50% for the oral exam. Problem solving capability, clarity of exposition and correctness of the information will be the main criteria of the evaluation.</i></p>
<b>Additional information</b>	